# organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 293 KMean  $\sigma$ (C–C) = 0.004 Å R factor = 0.030 wR factor = 0.078 Data-to-parameter ratio = 21.0

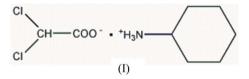
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# Cyclohexylammonium dichloroacetate

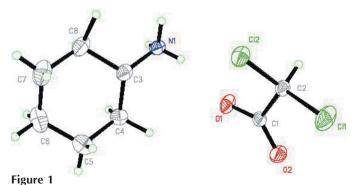
The title compound,  $C_6H_{14}N^+ \cdot C_2HCl_2O_2^-$ , (I), was obtained as a byproduct of the oxidation of the chiral compound 5-(*R*)menthyloxybutyrolacto[3,4-*b*]-2(*S*)-6(*R*)-1-*N*-cyclohexylaziridine. The structure determination of (I) shows that the cations and anions are connected through weak intermolecular hydrogen bonding, forming a complicated network structure.

### Comment

Weak non-covalent interactions between molecules, such as hydrogen bonds and  $\pi$ - $\pi$  stacking effects, can play important roles in determining crystal structures (Desiraju *et al.*, 1999; Mingos, 2004; Ma *et al.*, 2005; Zhu, *et al.*, 2004; Men, *et al.*, 2005; Dimartino *et al.*, 2005; Sontjens *et al.*, 2001; Kim, *et al.*, 2005). Recently, 5-(*R*)-menthyloxybutyrolacto[3,4-*b*]-2(*S*)-6(*R*)-1-*N*-cyclohexylaziridine was oxidized at room temperature when treated with hydrogen peroxide and hydrochloric acid (Guo *et al.*, 2005). However, instead of the expected aziridine product containing two carboxyl groups, the title compound, (I), was obtained. We report here its crystal structure.



The structure of (I) (Fig. 1) comprises a cyclohexylammonium cation and a dichloroacetate anion. The C1-O1 and C1-O2 bond lengths in the carboxylate group are 1.240 (2) and 1.232 (2) Å, respectively. Hydrogen-bonding interactions play an important role in the structure of the title compound (Fig. 2), with a number of different kinds of



The structure (30% probability displacement ellipsoids) of the title compound.

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hydrogen bonding, *viz*. hydrogen bonding between N–H and the O atoms of three dichloroacetate anions, and hydrogen bonding between N–H and the Cl atoms (Table 1). Noteworthy is the presence of a bifurcated hydrogen bond, N1–  $H1A\cdots O1$  and N1– $H1A\cdots Cl2$ . In addition, there are also a number of weak C– $H\cdots Cl$  and C– $H\cdots O$  hydrogen bonds in the structure.

## **Experimental**

5-(R)-Menthyloxybutyrolacto[3,4-b]-2(S)-6(R)-1-N-cyclohexylaziridine (0.671 g, 2 mmol) was dissolved in 5 ml acetonitrile. Concentrated hydrochloric acid (0.344 ml, 4 mmol) and excess hydrogen peroxide (2 ml) were then added and the mixture stirred for 24 h at room temperature. The crude product obtained was purified by column chromatography on silica gel to obtain the purified compound (0.411 g). Colourless crystals were grown by slow evaporation of a petroleum–ethanol solution (4:1 v/v).

Crystal data

$C_6H_{14}N^+ \cdot C_2HCl_2O_2^-$	Mo $K\alpha$ radiation
$M_r = 228.10$	Cell parameters from 5012
Orthorhombic, $P2_12_12_1$	reflections
a = 6.6435 (8) Å	$\theta = 2.6 - 28.6^{\circ}$
b = 8.5960 (10)Å	$\mu = 0.56 \text{ mm}^{-1}$
c = 19.135 (2) Å	T = 293 (2) K
V = 1092.8 (2) Å <sup>3</sup>	Block, colourless
Z = 4	$0.36 \times 0.18 \times 0.16 \ \mathrm{mm}$
$D_x = 1.380 \text{ Mg m}^{-3}$	

Data collection

Bruker SMART CCD area-detector	2308 reflections with $I > 2\sigma(I)$
diffractometer	$R_{\rm int} = 0.021$
$\varphi$ and $\omega$ scans	$\theta_{\rm max} = 27.5^{\circ}$
Absorption correction: none	$h = -8 \rightarrow 8$
9724 measured reflections	$k = -11 \rightarrow 11$
2497 independent reflections	$l = -23 \rightarrow 24$

## Refinement

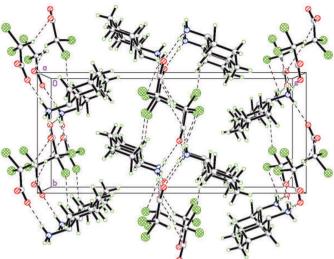
Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.0362P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.030$	+ 0.2076P]
$wR(F^2) = 0.078$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.08	$(\Delta/\sigma)_{\rm max} < 0.001$
2497 reflections	$\Delta \rho_{\rm max} = 0.28 \text{ e} \text{ Å}^{-3}$
119 parameters	$\Delta \rho_{\rm min} = -0.30 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	Absolute structure: Flack (1983),
	with 1020 Friedel pairs
	Flack parameter: 0.02 (7)

## Table 1

Hydrogen-bond geometry (Å, °).

$\overline{D - \mathbf{H} \cdots A}$	<i>D</i> -Н	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$ \begin{array}{c} N1 - H1C \cdots O1^{i} \\ N1 - H1B \cdots O2^{ii} \\ N1 - H1A \cdots Cl2 \\ N1 - H1A \cdots Cl2 \\ N1 - H1A \cdots O1 \end{array} $	0.89	1.94	2.8194 (19)	170
	0.89	1.89	2.7765 (19)	175
	0.89	2.90	3.3868 (15)	116
	0.89	1.95	2.8237 (19)	166

Symmetry codes: (i)  $x - \frac{1}{2}, -y + \frac{1}{2}, -z$ ; (ii) x - 1, y, z.



### Figure 2

View along the a axis of the packing of the title compound, showing the hydrogen-bonding interactions as dashed lines.

H atoms were placed in calculated positions (N–H = 0.89, C–H = 0.97 Å) and refined as riding, with  $U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm N})$  and  $1.2U_{\rm eq}({\rm C})$ .

Data collection: *APEX2* (Bruker 2004); cell refinement: *APEX2*; data reduction: *SAINT* (Bruker 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker 2004); software used to prepare material for publication: *SHELXTL*.

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## References

- Bruker (2004). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Desiraju, G. R. & Steiner, T. (1999). The Weak Hydrogen Bond In Structural Chemistry and Biology. IUCr Monographs on Crystallography 9. Oxford: Oxford Science Publications.
- Dimartino, G., Wang, D., Chapman, R. N. & Arora, P. S. (2005). Org. Lett. 7, 2389–2392.

Flack, H. D. (1983). Acta Cryst. A39, 876-881.

Guo, J. B., Li, S. L., Yu, Z. L., Chen, Q. H. (2005). Chin J. Struct. Chem. 24, 89– 93.

Kim, K. M., Park, H., Kim, H. J., Chin, J., & Nam, W. (2005). Org. Lett. 7, 3525– 3527.

Ma, L. F., Zhao, B. T., Wang, L. Y. (2005). Acta Cryst. E61, 0964-0966.

Men, Y. B., Liu, H., Zhang, X. Z. & Zhong, B. H. (2005). Anal. Sci. 21, x123x124.

Mingos, D. M. P. (2004). Supramolecular Assembly via Hydrogen Bonds II. Structure and Bonding, p.180. Berlin: Springer-Verlag.

Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

- Sheldrick, G. M. (1997). SHELXL97, University of Göttingen, Germany.
- Sontjens, S. H. M., Meijer, J. T., Kooijman, H., Spek, A. L., Genderen, M. H. P., Sijbesma, R. P., & Meijer, E. W. (2001). Org. Lett. 3, 3887–3889.
- Zhu, S., Xing, C., Xu, W., Jin, G., & Li, Z. (2004). Cryst. Growth Des. 4, 53-56.